

**4-MERCAPTO-5,5-DIMETHYL-2-PHENYL-
2,3,5,6-TETRAHYDRO-7H-THIOCHROMEN-
7-YLIDENEMORPHOLINIUM PERCHLORATE**

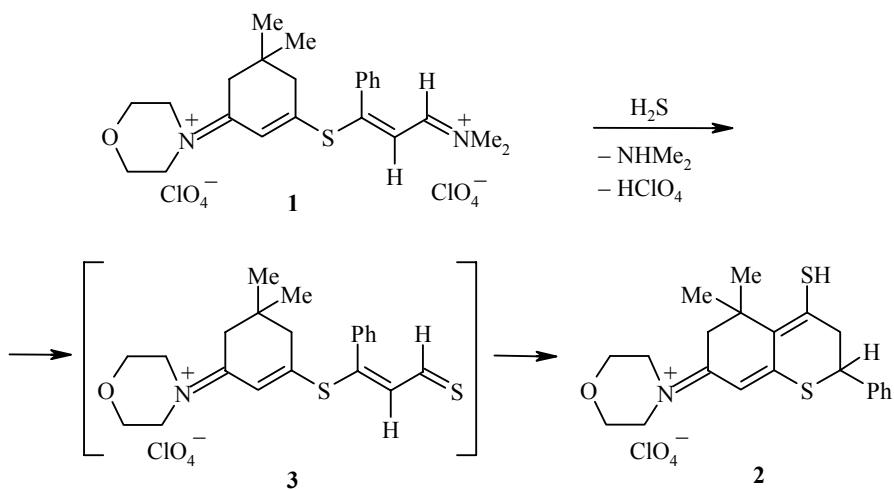
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Keywords: (1-N,N-dimethylimmonio-3-phenylprop-2-en-3-yl) (5,5-dimethyl-1-morpholinocyclohex-2-en-3-yl) sulfide diperchlorate, 4-mercaptop-5,5-dimethyl-2-phenyl-2,3,5,6-tetrahydro-7H-thiochromen-7-ylidenemorpholinium perchlorate, hydrothiolytic.

We showed earlier that (1-N,N-dimethylimmonio-3-phenylprop-2-en-3-yl) (5,5-dimethyl-1-morpholinocyclohex-2-en-3-yl) sulfide diperchlorate (**1**) in methanol medium in the presence of a catalytic amount of triethylamine is readily converted to 5,5-dimethyl-7-morpholinio-2-phenyl-5,6-dihydro-7H-thiochromene perchlorate [1].

While studying hydrothiolytic reaction of diperchlorate **1**, we found that the end product of its reaction with hydrogen sulfide in DMF is the previously unknown 4-mercaptop-5,5-dimethyl-2-phenyl-2,3,5,6-tetrahydro-7H-thiochromen-7-ylidenemorpholinium perchlorate (**2**). The process probably occurs *via* intermediate formation of (3-phenyl-1-thioxoprop-2-en-3-yl) (5,5-dimethyl-1-morpholinocyclohex-2-en-3-yl) sulfide (**3**), which then undergoes ring closure.

This reaction is a novel example of synthesis of difficultly accessible thiochromene derivatives starting from enamino thioketones and halosubstituted immonium salts [2].



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4-Mercapto-5,5-dimethyl-2-phenyl-2,3,5,6-tetrahydro-7H-thiochromen-7-ylidenemorpholinium Perchlorate (2). Dry hydrogen sulfide was passed through a solution of (1-N,N-dimethylimmonio-3-phenylprop-2-en-3-yl) (5,5-dimethyl-1-morpholinocyclohex-2-en-3-yl) sulfide diperchlorate (**1**) (0.24 g, 0.4 mmol) [2] in DMF (5 ml) at 20°C for 30 min (TLC monitoring, Silufol, chloroform–ethyl acetate, 3:1). The reaction mixture was poured into ice–water mixture (30 g). The precipitate was filtered out, washed with water, and dried under vacuum over CaCl_2 . Obtained 0.15 g (80%) of compound **2** as a mustard color powder with mp 123–125°C. ^1H NMR spectrum (400 MHz, DMSO-d_6 , HMDS), δ , ppm: 0.82 and 0.87 (6H, s, 2CH_3); 2.44 (2H, m, 6- CH_2); 2.64 (2H, m, 3- CH_2); 3.55 (1H, s, SH); 3.66 (1H, m, 2-CH); 3.67 and 3.83 (8H, m, $\text{CH}_2\text{N}+\text{CH}_2$ and CH_2OCH_2); 6.30 (1H, s, HC=); 7.35–7.57 (5H, m, C_6H_5). ^{13}C NMR spectrum (DMSO-d_6 , HMDS), δ , ppm: 27.4, 27.6 (CH_3); 34.5 ($\text{C}_{(5)}$); 43.0 ($=\text{N}^+(\text{CH}_3)_2$); 50.3 ($\text{C}_{(6)}$); 50.9 ($\text{C}_{(3)}$); 63.4 ($\text{O}(\text{CH}_2)_2$); 66.1 ($\text{C}_{(2)}$); 110.0 ($\text{C}=\text{N}^+$); 122.1 (=C–H); 127–140 (CPh, $\text{C}_{(4)}$, $\text{C}_{(9)}$, $\text{C}_{(10)}$). IR spectrum (KBr), ν , cm^{-1} : 623 and 1096 (ClO_4^-), 1556, 1576 ($\text{C}=\text{C}$), 1651 ($\text{C}=\text{N}^+$), 2484 (SH). Found, %: C 54.20; H 5.45; Cl 6.84; N 2.17; S 13.82. $\text{C}_{21}\text{H}_{26}\text{ClNO}_5\text{S}_2$. Calculated, %: C 53.45; H 5.51; Cl 7.53; N 2.97; S 13.57.

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